

IN THE CLAIMS:

This listing of claims will replace all prior versions, and listings, of claims in the application:

Listing of Claims:

Claims 1 - 94 (Cancelled).

95. (Currently Amended) A method for controlling microbial or biofilm growth in a medium, the method comprising mixing a nitrogen-containing compound or a mixture of such compounds, said nitrogen-containing compound being a salt containing nitrogen both in the cation portion and in the anion portion thereof, selected from the group consisting of ~~of~~ salts of the formula $Y^{x-} [NH_2R^3R^4]Z^{n+}$, wherein x is 1 to 3, Y^{x-} is a basic form of an acid Y that contains at least one moiety selected from the group consisting of a primary amine moiety, a secondary amine moiety, ~~a tertiary amine moiety~~, an amide moiety, an imide moiety, a sulfamide moiety, a sulfimide moiety, and an amineimine moiety, and Z^{n+} ~~is a cation other than a cation of the form $[NH_2R^3R^4]^+$ wherein $[NH_2R^3R^4]^+$ is an acidic form of a base NHR^3R^4 wherein R^3 and R^4 are each independently selected from the group consisting of H and C_{1-8} alkyl, or R^3 and R^4 , together with the nitrogen atom to which they are attached, form a 5- to 10-member heterocyclic ring optionally~~

substituted by one or more groups selected from C₁₋₆ alkyl, C₃₋₈ cycloalkyl, halogen, hydroxy, -OC₁₋₆ alkyl or -OC₃₋₈ cycloalkyl, ~~and n is a whole number greater than zero; and~~

~~(ii) amphoteric molecules Q containing at least one moiety selected from the group consisting of COOH and SO₃H and at least one moiety selected from the group consisting of a primary amine moiety, a secondary amine moiety, and a tertiary amine moiety;~~

and an aqueous solution of a hypochlorite oxidant to form a biocide, wherein the molar ratio of [NH₂R³R⁴]⁺ nitrogen atoms ~~in said salt to~~ said hypochlorite is at least 1:1, and applying said biocide to said medium.

Claim 96 (Cancelled).

97. (Currently Amended) A method according to claim ~~96~~95, wherein Y^{x-} is of the formula [R¹R²HN-A-COO]^{x-} or [R¹R²HN-A-SO₃]^{x-}, wherein:

A is a bond, straight-chain or branched C₂₋₂₀ alkyl, straight-chain or branched C₂₋₂₀ alkenyl, straight-chain or branched C₂₋₂₀ alkynyl, C₃₋₁₀ cycloalkyl, straight-chain or branched C₄₋₁₀ cycloalkenyl, C₄₋₁₀ cycloalkynyl, or C₆₋

C₁₀ aryl, wherein each C₁₋₂₀ alkyl, C₂₋₂₀ alkenyl, C₂₋₂₀ alkynyl, C₃₋₁₀ cycloalkyl, C₄₋₂₀ alkylcycloalkyl, C₄₋₁₀ cycloalkenyl, C₄₋₁₀ cycloalkynyl or C₆₋₁₀ aryl is optionally substituted with one or more groups selected from -COOH, -COH, -SCH₃, -NH₂, =NH, -NHC(=NH)NH₂, -C(=O)NH₂, -OH, 4-hydroxyphenyl, 5-imidazolyl, 3-indolyl, halogen, -SO₃H, =O, C₁₋₈ alkyl, C₃₋₈ cycloalkyl, C₄₋₉ cycloalkylalkyl, phenyl, 4-methylphenyl, benzyl, ~~-O-C₃₋₈ cycloalkyl~~, ~~-O-C₃₋₈ cycloalkyl~~, -O-C₃₋₈ cycloalkyl, -O-C₄₋₉ cycloalkylalkyl, -O-phenyl, -O-4-methylphenyl, -O-benzyl, -SO₂R⁷ or -NHR⁷ wherein R⁷ is H, C₁₋₈ alkyl, phenyl, 4-methylphenyl, benzyl or -NH₂, and wherein each C₁₋₂₀ alkyl, C₂₋₂₀ alkenyl, C₂₋₂₀ alkynyl, C₃₋₁₀ cycloalkyl, C₄₋₂₀ alkylcycloalkyl, C₄₋₁₀ cycloalkenyl, C₄₋₁₀ cycloalkynyl or C₆₋₁₀ aryl optionally contains one to three heteroatoms selected from N, O and S;

~~R¹ and R² are~~ is each independently selected from the group consisting of H, straight-chain or branched C₁₋₂₀ alkyl, straight-chain or branched C₂₋₂₀ alkenyl, straight-chain or branched C₂₋₂₀ alkynyl, C₃₋₁₀ cycloalkyl, straight-chain or branched C₄₋₂₀ alkylcycloalkyl, C₄₋₁₀ cycloalkenyl, C₄₋₁₀ cycloalkynyl, or C₆₋₁₀ aryl, wherein each C₁₋₂₀ alkyl, C₂₋₂₀ alkenyl, C₂₋₂₀ alkynyl, C₃₋₁₀ cycloalkyl, C₄₋₂₀ alkylcycloalkyl, C₄₋₁₀ cycloalkenyl, C₄₋₁₀

cycloalkynyl or C₆-C₁₀ aryl is optionally substituted with one or more groups selected from -COOH, -COH, -SCH₃, -NH₂, =NH, -NHC(=NH)NH₂, -C(=O)NH₂, -OH, 4-hydroxyphenyl, 5-imidazolyl, 3-indolyl, halogen, -SO₃H, =O, C₁₋₈ alkyl, C₃₋₈ cycloalkyl, C₄₋₉ cycloalkylalkyl, phenyl, 4-methylphenyl, benzyl, -O-C₂₋₈ cycloalkyl, -O-C₃₋₈ cycloalkyl, -O-C₄₋₉ cycloalkylalkyl, -O-phenyl, -O-4-methylphenyl, -O-benzyl, -SO₂R⁷ or -NHR⁷ wherein R⁷ is H, C₁₋₈ alkyl, phenyl, 4-methylphenyl, benzyl or -NH₂, and wherein each C₁₋₂₀ alkyl, C₂₋₂₀ alkenyl, C₂₋₂₀ alkynyl, C₃₋₁₀ cycloalkyl, C₄₋₂₀ alkylcycloalkyl, C₄₋₁₀ cycloalkenyl, C₄₋₁₀ cycloalkynyl or C₆-C₁₀ aryl optionally contains one to three heteroatoms selected from N, O and S;

or R¹ and A, together with the nitrogen atom to which they are attached, form a 5- to 10-member heterocyclic ring or a 5- to 10-member heteroaromatic ring in which the free electron pair of the nitrogen atom to which R¹ and A is attached is not part of the aromatic pi-electron system, the 5- to 10-member heterocyclic or heteroaromatic ring being optionally substituted by one or more groups selected from C₁₋₆ alkyl, C₃₋₈ cycloalkyl, halogen, hydroxy, -OC₁₋₆ alkyl or -OC₃₋₈ cycloalkyl.

~~or R^1 and R^2 , together with the nitrogen atom to which they are attached, form a 5 to 10 member heterocyclic ring or a 5 to 10 member heteroaromatic ring in which the free electron pair of the nitrogen atom to which R^1 and A is attached is not part of the aromatic pi electron system, the 5 to 10 member heterocyclic or heteroaromatic ring being optionally substituted by one or more groups selected from C_{1-6} alkyl, C_{3-8} cycloalkyl, halogen, hydroxy, OC_{1-6} alkyl or OC_{3-8} cycloalkyl.~~

Claim 98 (Cancelled).

99. (Previously Presented) A method according to claim 95, wherein the concentration of said hypochlorite oxidant in said aqueous hypochlorite oxidant solution immediately prior to mixing with said nitrogen-containing compound is not more than 24,000 ppm as total chlorine.

100. (Previously Presented) A method according to claim 95, wherein said nitrogen-containing compound or mixture thereof is in an aqueous solution at a concentration of 0.5-60% w/v prior to mixing with the hypochlorite oxidant solution.

101. (Previously Presented) A method according to claim 95, wherein said mixing takes place in a mixing chamber

into and out of which there is a continuous flow of water during said mixing.

102. (Previously Presented) A method according to claim 95, wherein said hypochlorite oxidant is selected from the group consisting of alkaline and alkali earth metal hypochlorites, hypochlorite released to water from a stable chlorine carrier and hypochlorite formed *in situ* from chlorine gas, and mixtures thereof.

103. (Previously Presented) A method according to claim 95, wherein said hypochlorite oxidant is selected from the group consisting of lithium hypochlorite, sodium hypochlorite, calcium hypochlorite, magnesium hypochlorite and potassium hypochlorite.

Claim 104 (Cancelled).

105. (Previously Presented) A method according to claim 95, wherein Y is selected from the group consisting of carbamic acid, sulfamic acid, glycine, glutamine, arginine, histidine, and lysine.

106. (Previously Presented) A method according to claim 101, wherein the concentration of said hypochlorite

oxidant in said aqueous hypochlorite oxidant solution prior to mixing with said nitrogen-containing compound is not more than 24,000 ppm as total chlorine, and said mixing chamber comprises a conduit through which water flows as said hypochlorite oxidant solution and the nitrogen-containing compound are mixed.

107. (Previously Presented) A method according to claim 106, wherein said solution of hypochlorite oxidant is prepared *in situ* in said conduit prior to addition of said solution of said nitrogen-containing compound to said conduit.

108. (Previously Presented) A method according to claim 95, wherein said nitrogen-containing compound is diluted prior to mixing with the hypochlorite oxidant.

109. (Previously Presented) A method according to claim 95, wherein said medium is pulp and paper factory process water.

110. (Previously Presented) A method according to claim 95, wherein said medium is cooling tower water.

111. (Previously Presented) A method according to claim 95, wherein said medium is waste water or reclaimed waste water.

112. (Previously Presented) A method according to claim 95, wherein said medium is a clay slurry.

113. (Previously Presented) A method according to claim 95, wherein said medium is a starch slurry.

114. (Previously Presented) A method according to claim 95, wherein said medium is a sludge.

115. (Previously Presented) A method according to claim 95, wherein said medium is soil.

116. (Previously Presented) A method according to claim 95, wherein said medium is a colloidal suspension.

117. (Previously Presented) A method according to claim 95, wherein said medium is irrigation water.

118. (Previously Presented) A method according to claim 95, wherein said medium is a medium containing strong reducing agents.

119. (Previously Presented) A method according to claim 95, wherein said medium is a medium having a high reducing capacity.

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Claim 120 (Cancelled).

Claim 121 (Cancelled).

122. (Previously Presented) A method according to claim 95, wherein the concentration of said biocide immediately prior to being applied to said medium is from 1000 to 12,000 ppm expressed as total chlorine.

123. (Previously Presented) A method according to claim 95, wherein the concentration of said biocide in said medium, upon application of the biocide to said medium, is 0.5-300 ppm expressed as chlorine.

124. (Previously Presented) A method according to claim 95, wherein said biocide is effective within 1 hour of application to said medium.

125. (Withdrawn - Currently Amended) Apparatus for applying a biocide to a medium, comprising:

a nitrogen-containing compound reservoir containing a nitrogen-containing compound or mixture thereof selected from the group consisting of:

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salts of the formula $Y^{x-} [NH_2R^3R^4] Z^{n+}_{x/n}$, wherein x is 1 to 3, Y^{x-} is a basic form of an acid Y that contains at least one moiety selected from the group consisting of a primary amine moiety, a secondary amine moiety, ~~a tertiary amine moiety,~~ an amide moiety, an imide moiety, a sulfamide moiety, a sulfimide moiety, and an amineimine moiety, and ~~Z^{n+} is a cation other than a cation of the form $[NH_2R^3R^4]^+$ wherein $[NH_2R^3R^4]^+$ is an acidic form of a base NHR^3R^4 wherein R^3 and R^4 are each independently selected from the group consisting of H and C_{1-8} alkyl, or R^3 and R^4 , together with the nitrogen atom to which they are attached, form a 5- to 10-member heterocyclic ring optionally substituted by one or more groups selected from C_{1-6} alkyl, C_{3-8} cycloalkyl, halogen, hydroxy, $-OC_{1-6}$ alkyl or $-OC_{3-8}$ cycloalkyl, and n is a whole number greater than zero; and~~

~~amphoteric molecules Q containing at least one moiety selected from the group consisting of COOH and SO_3H and at least one moiety selected from the group consisting of a primary amine moiety, a secondary amine moiety, and a tertiary amine moiety;~~

a source of hypochlorite oxidant dilution having a concentration of between not more than 24,000 ppm as total chlorine,

and a mixing chamber operable to mix the dilution and the nitrogen-containing compound or mixture thereof in a molar ratio of nitrogen atoms in the nitrogen-containing compound to the hypochlorite of at least 1:1, to produce the biocide in the mixing chamber.

Claim 126 (Cancelled).

127. (Withdrawn) Apparatus according to claim 125, wherein said source of hypochlorite oxidant dilution comprises a hypochlorite-containing reservoir containing a hypochlorite oxidant solution, and a diluter operable to dilute the hypochlorite oxidant solution to produce said hypochlorite oxidant dilution having a concentration of not more than 24,000 ppm expressed as total chlorine.

128. (Withdrawn) Apparatus according to claim 127, wherein said diluter and said mixing chamber are a single conduit which is adapted to dilute said hypochlorite oxidant prior to mixing with said nitrogen-containing compound or mixture thereof.

129. (Currently Amended) A method for controlling microbial or biofilm growth in a medium, the method comprising

mixing a nitrogen-containing compound, a bromide and an aqueous solution of a hypochlorite oxidant to form a biocide, said nitrogen-containing compound being ~~selected from the group consisting of~~ a salts of the formula $Y^{x-}[NH_2R^3R^4]^+_x$, salts of the formula $Y^{x-}Z^{n+}_{x/n}$, and molecules Y per se, containing nitrogen both in the cation portion and in the anion portion thereof, wherein

~~Z^{n+} is a cation other than a cation of the form $[NH_2R^3R^4]^+_1$ wherein $[NH_2R^3R^4]^+_1$ is as defined below, and n is a whole number greater than zero~~

Y^{x-} is a basic form of an acid Y that contains at least one moiety selected from the group consisting of a primary amine moiety, a secondary amine moiety, ~~a tertiary amine moiety~~, an amide moiety, an imide moiety, a sulfamide moiety, a sulfimide moiety, and an amineimine moiety; and

$[NH_2R^3R^4]^+$ is an acidic form of a base NHR^3R^4 wherein:

R^3 and R^4 are each independently selected from the group consisting of H and C_{1-8} alkyl, or R^3 and R^4 , together with the nitrogen atom to which they are attached, form a 5- to 10-member heterocyclic ring optionally substituted by one or more

groups selected from C₁₋₆ alkyl, C₃₋₈ cycloalkyl, halogen, hydroxy, -OC₁₋₆ alkyl or -OC₃₋₈ cycloalkyl; and

x is 1 to 3;

and the molar ratio of ~~nitrogen atoms in said~~
~~nitrogen-containing compound~~ $[\text{NH}_2\text{R}^3\text{R}^4]^+$ to hypochlorite is at least 1:1,

and applying said biocide to said medium.

130. (New) A method according to claim 95, wherein said nitrogen-containing compound is ammonium carbamate or ammonium sulfamate.

131. (New) A method according to claim 95, wherein said nitrogen-containing compound is ammonium carbamate.

132. (New) A method according to claim 95, wherein said hypochlorite oxidant is sodium hypochlorite.

133. (New) A method according to claim 95, wherein said hypochlorite oxidant is sodium hypochlorite, said nitrogen-containing compound is ammonium carbamate and said medium is waste water or reclaimed waste water.